

(1)

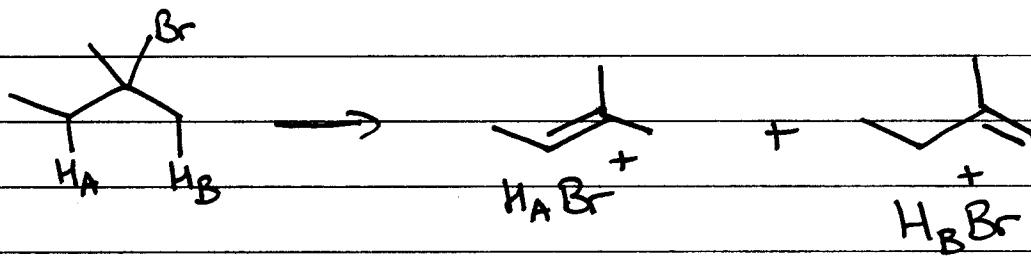
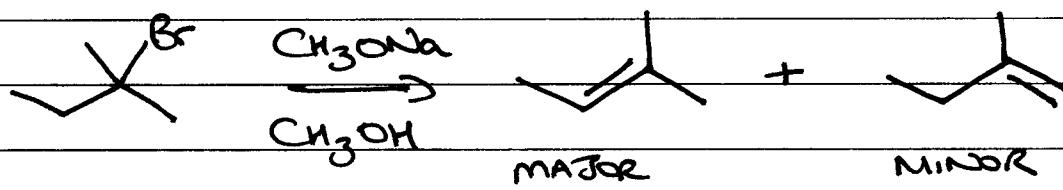
LEC 22

CHEM 30A

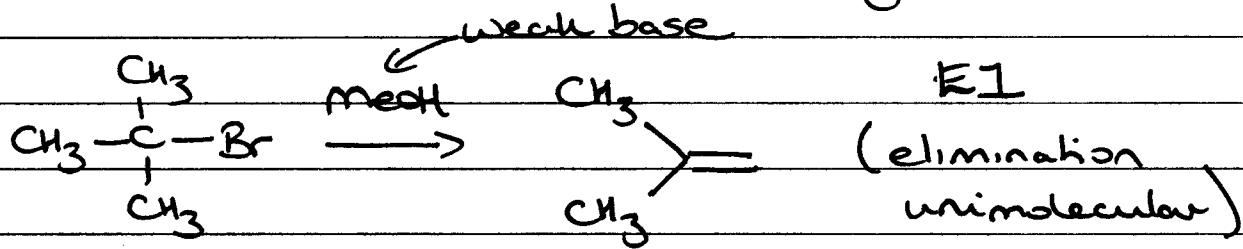
Nov 30<sup>th</sup>

- (1) INTRO TO  $\beta$ -ELIMINATION
- (2) MECHANISMS
- (3) STEREOCHEMISTRY
- (4) SUMMARY

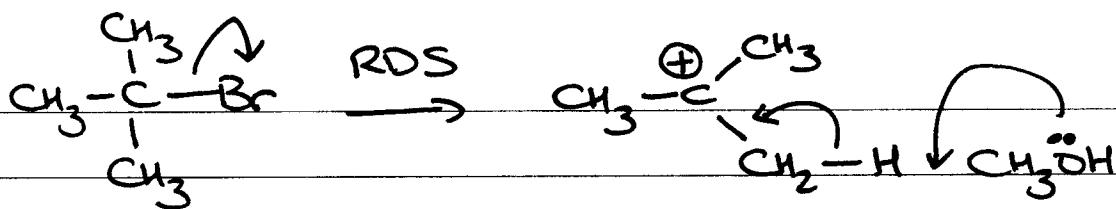
READ 9.5 - 9.11, PROBLEMS 9.37 - 9.42

(1)  $\beta$ -ELIMINATIONZAITSEV'S RULE  $\rightarrow$  major product is the most substituted alkene (more STABLE)

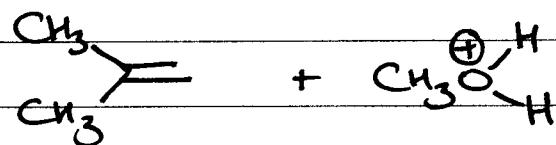
## (2) MECHANISMS

(like S<sub>N</sub> reactions, two limiting ones)

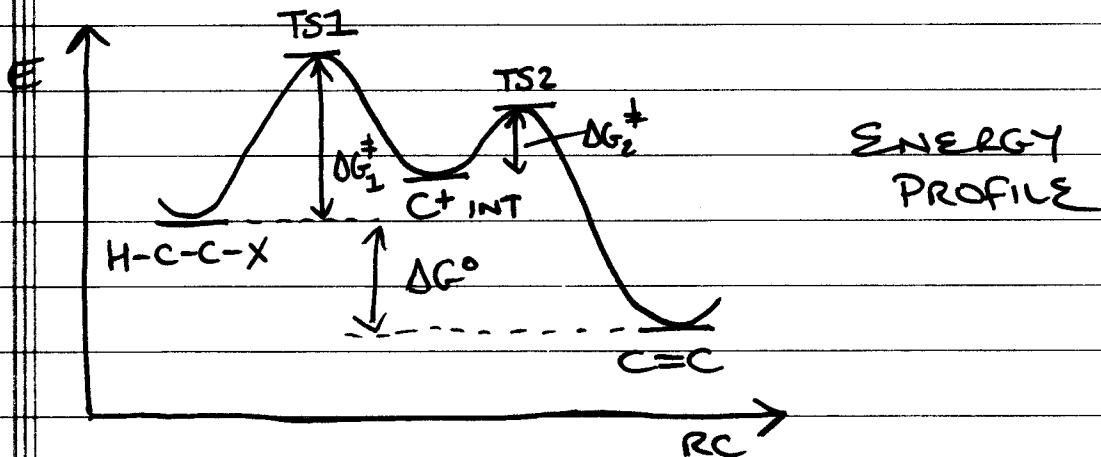
(2)



Competes with  
S<sub>N</sub>1 reaction



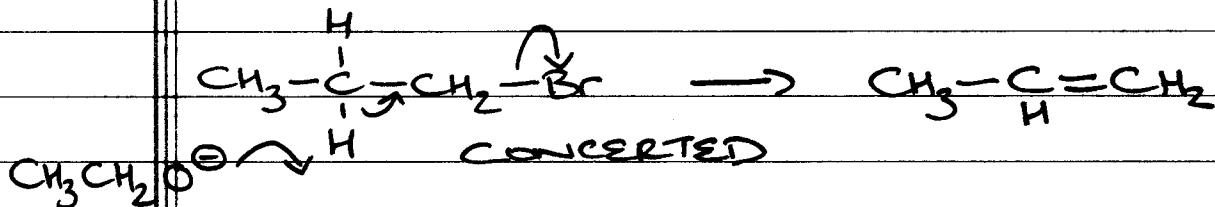
$$\text{rate} = k_1 [(\text{CH}_3)_3\text{C}-\text{Br}]$$



## E2 (ELIMINATION BIMOLECULAR)

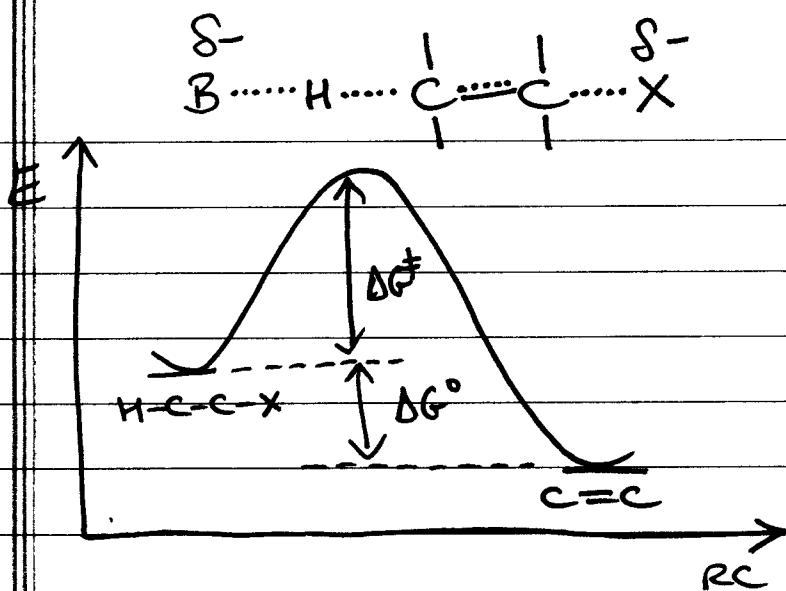


(competes with S<sub>N</sub>2)



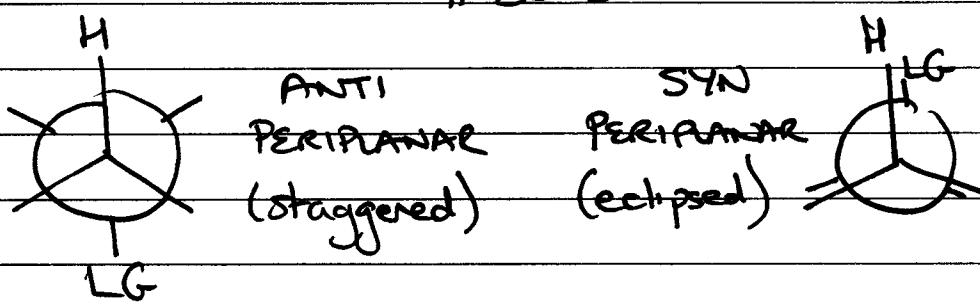
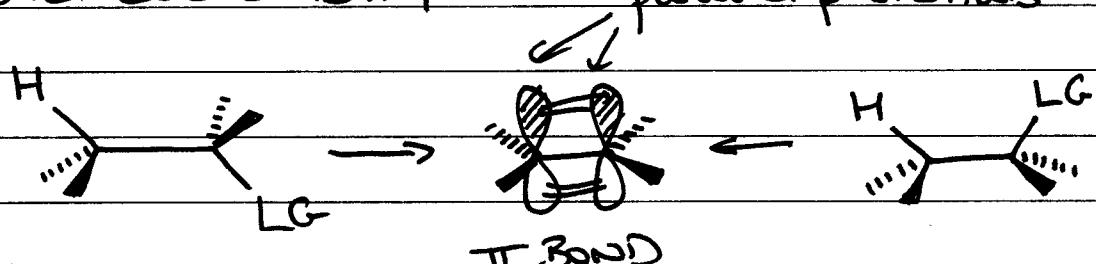
$$\text{rate} = k_2 [\sim \text{Br}][\text{Base}]$$

(3)

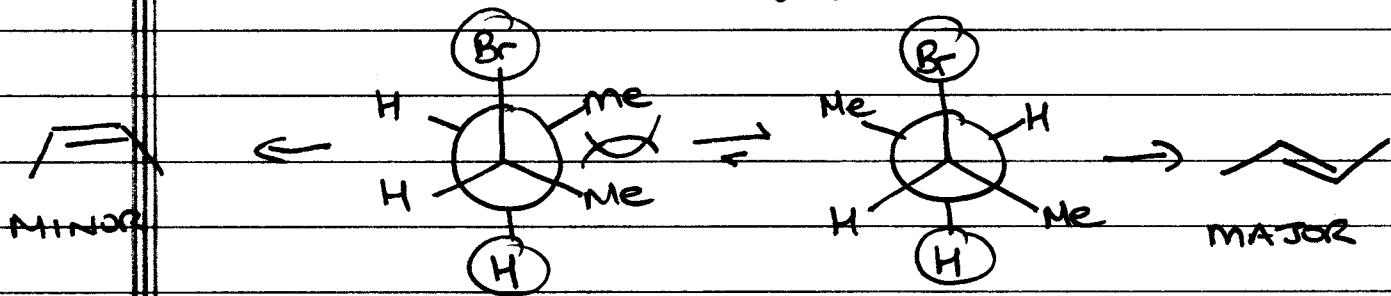
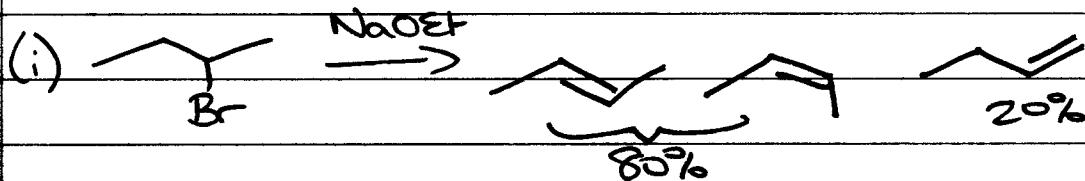


## (3) STEREOCHEMISTRY

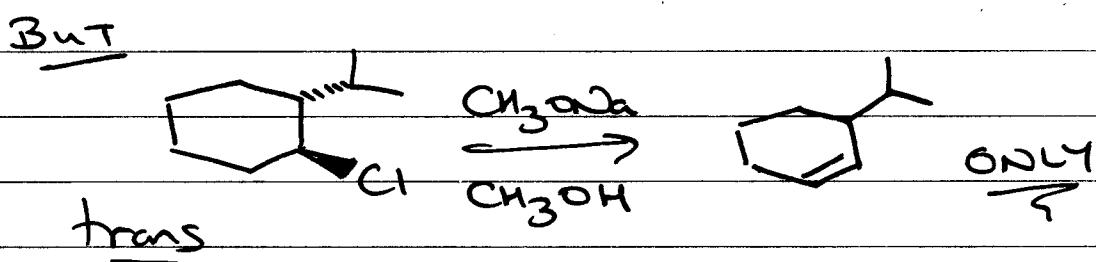
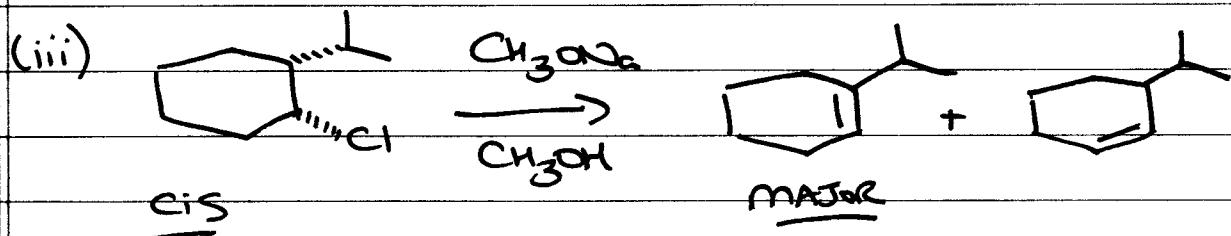
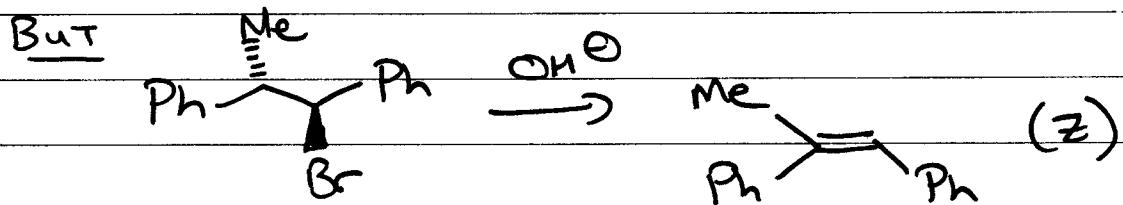
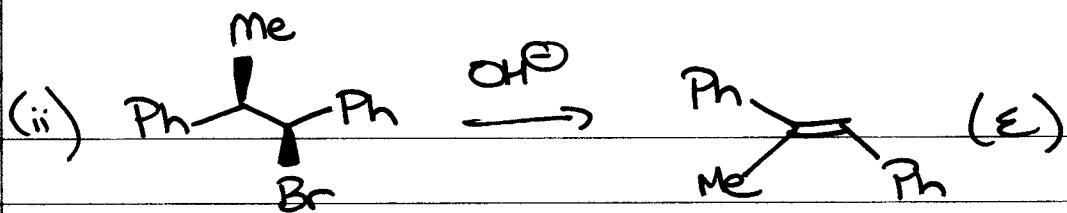
parallel p orbitals



Generally, anti-periplanar geometry is preferred in an E2 reaction (exceptions)



4



also, cis reacts faster  $\Rightarrow$  why?

