PREPARATION OF DI- AND TRIACYLIMINES AND THEIR USE IN THE SYNTHESIS OF NITROGEN HETEROCYCLES

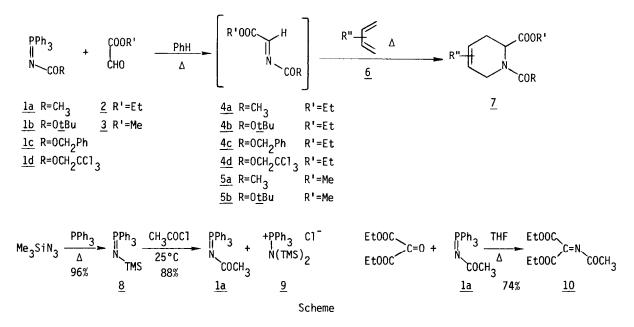
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<u>Abstract</u>: Reaction of aza-Wittig reagents with glyoxylates and keto malonates produces di- and tri acylimines which are moderately reactive dienophiles for Diels-Alder cycloaddition.

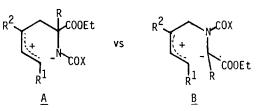
In the last few years, a considerable amount of synthetic effort has been expended on demonstrat ing the utility of the imino Diels-Alder reaction.² In the course of work directed toward the tota synthesis of pyrrolizidine and indolizidine alkaloids, we have investigated the synthesis of di- an triacylimines and their reactivity as Diels-Alder dienophiles.

The diacylimines 4a-d and 5ab were prepared (Scheme) by an aza-Wittig reaction between the phosphineimines <u>la-d</u> and the glyoxylates <u>2</u> or <u>3</u>.³ These diacylimines were not isolated but rather reacted <u>in situ</u> with various dienes <u>6</u> to give fair to good yields of the corresponding cycloadducts <u>7</u>. The N-acetylphosphineimine <u>la</u> has been reported previously from the reaction of iminotriphenylphosphorane and acetyl chloride. However, to avoid the use of the readily hydrolyzable N-unsubstitute compound, ⁵ the preparation of <u>la</u> from N-trimethylsilyliminotriphenylphosphorane <u>8</u> was effected. Reaction of trimethylsilyl azide with triphenylphosphine afforded a 96% yield of <u>8</u>⁶ which was reacted with acetyl chloride to produce <u>la</u> and <u>9</u> in 88% yield. The N-carboalkoxyiminotriphenylphosphoranes <u>1b-d</u> were all prepared from the corresponding azides.⁷ The triacylimine <u>10</u>, prepared by a similar approach from the aza-Wittig reagent <u>la</u> and diethyl oxomalonate⁸ in 74% yield, could be isolated as stable oil (bp 105-110°C, 0.25 Torr).

The Diels-Alder reactions were generally carried out in refluxing benzene or dimethoxyethane (DM for 15 hours. The results of these cycloadditions are shown in the Table. Several observations should be made. The di- and triacylimines - 4, 5, and 10 - are moderately active dienophiles, givin good yields of adducts only with activated dienes [e.g., Danishefsky's diene $\underline{6a}$, 9a Cohen's diene $\underline{6b}$, 9b and cyclopentadiene $\underline{6e}$. With less reactive dienes, e.g., $\underline{6c}^{9c}$ and $\underline{6d}^{9d}$ the yields are only fair. It is to be noted that all of these cycloadditions are carried out under one atmosphere of pressure and did not require the high pressure conditions described by Plieninger.³ Reaction of the imine $\underline{4b}$ with unsymmetrical dienes having a substituent in the 1-position, i.e., $\underline{6a}$ and $\underline{6b}$, proceeding regiospecifically with the substituent ending up α to the nitrogen atom in the product, i.e., $\underline{7a}$ and $\underline{7b}$. This is in agreement with the generalization of Weinreb^{2b} concerning the regiochemistry of imine Diels-Alder reactions. Thus one would expect the major product to be produced via the more stable



polar transition state (or intermediate) <u>A</u>, in which the negative charge rests on the activated nitrogen atom, rather than <u>B</u>. When the R group is an ester (so that the anion of <u>B</u> is an α -acet-amido malonate), namely in the triacylimine reaction (example 9), the energy difference is less and a 3:1 mixture of products, <u>7i</u> and <u>7j</u>, is formed with that resulting from <u>A</u>, <u>7i</u>, being predominant.



Chromatography of the mixture resulting from the reaction of $\underline{5b}$ with $\underline{6c}$ followed by methanolysis afforded a fair yield of the 4-piperidinone $\underline{7c}$ as the major product. We were unable to identify any of the corresponding regionsomer, the 3-piperidinone, among the minor byproducts, although we can not conclusively rule out its presence.

The structures of the adducts were inferred from their NMR, IR, and mass spectral data. For example, the ¹H NMR of the piperinone <u>7a</u> exhibited two doublets (J = 8 Hz) for the vinylogous amide protons at δ 7.85 and δ 5.30, thus implying that the cycloaddition had occurred with the assigned regiochemistry. The structure of <u>7c</u> was verified by comparison with an authentic sample prepared in 93% yield by catalytic hydrogenation of the methyl ester corresponding to <u>7a</u> (cycloaddition of <u>5a</u> and <u>6a</u>, 56%). The 200 MHz NMR spectrum of <u>7e</u> at 30°C exhibited two sets of peaks for each set of protons in the molecule. However, when the spectrum was taken at 100°C, each pair of peaks coalesced, thus indicating isomerism about the C-N amide linkage and not an <u>endo-exo</u> mixture. The <u>exo</u> stereochemistry of the ethyl ester was inferred from the fact that the proton α to the ester occurs as a clean singlet and thus should occupy the <u>endo</u> position. This result is not surprising for the following reasons. First, the two large groups should be <u>trans</u> in the dienophile <u>4a</u>. Second, since the

Example	Imine	Diene	Adduct	Yield(%)	Conditions
(1)	EtOOCCH=NCOO <u>t</u> Bu <u>4b</u>	TMS0	<u>7a</u> 0 COOEt N COO <u>t</u> Bu	84	1) PhH/∆/15h 2) H ₃ 0+
(2)	EtOOCCH=NCOO <u>t</u> Bu <u>4b</u>	Me0 SPh	$\frac{7b}{Me0} \xrightarrow{N} \frac{C00Et}{SPh}$	56	PhH/∆/12h
(3)	MeOOCCH=NCOO <u>t</u> Bu <u>5b</u>	TMS0 <u><u>6</u>c</u>	<u>7c</u> N COO <u>t</u> Bu	30	1) PhH/∆/15h 2) MeOH/K ₂ CO ₃
(4)	MeOOCCH=NCOO <u>t</u> Bu <u>5b</u>	Me0 <u>6d</u> Me0	<u>7d</u> Me0 C00Me Me0 C00 <u>t</u> Bu	16	PhH/∆/15h
	Et00CCH=NCOR	\square	CODE t		
(5) (6) (7) (8)	$\frac{4a}{4b} R = Me$ $\frac{4b}{4c} R = 0tBu$ $\frac{4c}{4c} R = 0CH_2Ph$ $\frac{4d}{4d} R = 0CH_2CC1_3$	<u>6e</u>	$\frac{7e}{7f} \qquad R = Me$ $\frac{7f}{7g} \qquad R = 0tBu$ $\frac{7g}{7h} \qquad R = 0CH_2Ph$ $\frac{7h}{7h} \qquad R = 0CH_2CC1_3$	81 55 57 61	DME/∆/15h
(9)	(EtOOC) ₂ C=NCOMe <u>10</u>	TMS0 <u>6a</u> OMe	COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt COOEt	45:15	1) 140°C/2h 2) H ₃ 0+

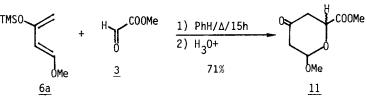
TABLE: Diels-Alder Reactions of Di- and Triacylimines

nitrogen dominates the orientational preference, the carbonyl group on nitrogen should be <u>endo</u> in the preferred transition state. Thus one would expect the formation of the adduct with the carbonyl group on nitrogen initially <u>endo</u> and the ethyl ester <u>exo</u>.

Finally, as a model for a Diels-Alder approach to glycosides,¹⁰ a mixture of $\underline{3}$ and $\underline{6a}$ in benzene was refluxed for 15h and then treated with 0.1N HCl in aqueous THF for 1h to produce in 71% yield an approximately 1:1 mixture of stereoisomers of $\underline{11}$.

The tetrahydropyridines and bicyclic amides produced by this methodology are potential intermediates for the construction of a variety of alkaloids. Further investigations along these lines are currently underway.

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- 9. The dienes were prepared by known methods as follows: a) 1-methoxy-3-trimethylsilyloxy-1,3-butadiene: S. Danishefsky and T. Kitahara, <u>J. Am. Chem. Soc.</u>, <u>96</u>, 7807 (1974); b) 2-methoxy-1-pheny thio-1,3-butadiene: T. Cohen, R. J. Ruffner, D. W. Shull, E. R. Fogel, and J. R. Falck, <u>Org. Synth.</u>, <u>59</u>, 202 (1979); c) 2-trimethylsilyloxy-1,3-butadiene: M. E. Jung and C. A. McCombs, <u>ibid.</u>, <u>58</u>, 163 (1978); d) 2,3-dimethoxy-1,3-butadiene: E. McDonald, A. Suksamrarn, and R. D. Wylie, <u>J. Chem. Soc. Perkin Trans. I</u>, 1893 (1979).
- For other similar approaches, see: a) S. David, <u>et al.</u>, <u>J. Chem. Soc. Perkin Trans. I</u>, 1795 (1979); 2274 (1974); <u>Chem. Commun.</u>, 535 (1978); <u>Tetrahedron</u>, 34, 299 (1978) and references therein. b) J. Jurczak, <u>et al.</u>, <u>J. Org. Chem.</u>, <u>44</u>, 3347 (1979) and references therein.

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