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Controlling the PCBM:Cathode Interface**

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Improving the Reproducibility of P3HT:PCBM Solar Cells by Controlling the PCBM:Cathode Interface

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Abstract

Plastic photovoltaic devices offer real potential for making solar energy economically viable. Unfortunately, bulk heterojunction (BHJ) solar cells fabricated from blends of the commonly-used materials poly(3-hexylthiophene), P3HT, and phenyl-C₆₁-butyric acid methyl ester, PCBM, sometimes exhibit low efficiencies even when the procedures followed often produce solar cells with efficiencies exceeding 5%. In this Letter, we show that this irreproducibility is caused by subtleties in the film processing conditions that ultimately lead to poor electron extraction from the devices. For low-performing devices, photogeneration and charge extraction with a linearly-increasing voltage ramp (photo-CELIV) measurements show an order-of-magnitude difference in the effective mobilities of the electrons and holes. Atomic force microscopy (AFM) experiments reveal that the top surface of these low-performing devices is nearly pure P3HT. We argue that small variations in the solvent evaporation kinetics during spin-coating of the BHJ active layer, which are difficult to control, cause PCBM to segregate toward the bottom of the P3HT film to different extents, explaining why electron extraction from the PCBM component of the BHJ is so difficult in poorly-performing devices. Finally, we show that electron extraction can be greatly improved by spin-coating a thin PCBM layer on top of the BHJ before deposition of the cathode, allowing the reproducible fabrication of high-efficiency polymer solar cells.

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3 Organic photovoltaics based on bulk-heterojunction (BHJ) composites of conjugated
4 polymers and fullerenes have shown rapid improvement in the last few years,[1, 2] with
5 power conversion efficiencies recently surpassing 6%.[3] Although facile, solution-phase fab-
6 rication is one of the greatest advantages this class of solar cells has over its inorganic-based
7 counterparts, the behavior of polymer:fullerene devices is sensitive to small variations in
8 processing conditions:[4, 5] *e.g.*, small changes in material blend ratios,[6] single-percent
9 variations in the composition of the solvent used for spin-coating,[7, 8] and changes in post-
10 fabrication treatments such as the time and/or temperature of thermal annealing[9] all can
11 dramatically affect device performance. Perhaps even more troublesome, there is not always
12 good reproducibility when different groups use the same processing recipe for producing
13 polymer:fullerene thin-film photovoltaic devices, indicating that there are still processing
14 parameters that we either have not yet correctly identified or properly learned to control in
15 order to consistently optimize device performance.

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17 A prime example of this lack of reproducibility can be seen in BHJ devices fabricated
18 from blends of the commonly-used materials regioregular poly(3-hexylthiophene), P3HT,
19 and phenyl-C₆₁ butyric acid methyl ester, PCBM. BHJ solar cells fabricated from these
20 materials can have power conversion efficiencies (PCEs) exceeding 5%,[9, 10] but sometimes
21 cells fabricated with nominally identical processing conditions can exhibit efficiencies less
22 than 1%. The low-performing devices often have an S-shaped current versus applied voltage
23 (J - V) behavior under illumination that results in exceedingly low fill-factors (\sim 10-25%).
24 Figure 1 shows J - V curves observed in our laboratory for P3HT:PCBM-based BHJ solar cells
25 fabricated using nominally identical processing conditions. The solid curve shows ‘typical’
26 J - V characteristics with a high fill-factor and good PCE. The dashed curve, on the other
27 hand, shows a device with a similar short-circuit current (J_{sc}) and open-circuit voltage (V_{oc})
28 but a fill-factor that is significantly reduced by the presence of the ‘S-curve’. Anecdotal
29 evidence suggests that few groups, if any, can reproducibly make \geq 5%-efficient devices
30 every time they try, and that nearly every group in the field fabricates devices that show
31 the S-curve from time to time. The sporadic nature of the S-curve, however, has prevented
32 widespread discussion about it in the literature: we are aware of only a few papers that have
33 investigated the origins of the S-curve in polymer:fullerene BHJ solar cells.[11–15]

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35 In this Letter, we identify the cause of the S-curve in P3HT:PCBM BHJ devices and
36 provide a method to reproducibly eliminate the low fill-factor through simple changes in the
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3 device architecture. Using the transient current technique known as ‘photogeneration and
4 charge extraction with a linearly-increasing voltage ramp’ (photo-CELIV),[16] we first show
5 that the S-curve results from poor contact between the PCBM component of the BHJ and the
6 cathode, which hinders electron extraction and leads to an imbalance in the rates at which
7 holes and electrons are extracted from the active layer. We then argue based on atomic force
8 microscopy (AFM) experiments that the poor extraction of charge at the cathode interface
9 is due to undesirable vertical phase segregation in the BHJ blend, the result of solvent
10 evaporation kinetics that can cause PCBM to segregate to the bottom of the film[17, 18]
11 to different extents depending on subtle variations in the BHJ film processing conditions.
12 Finally, we show that the S-curve can be reproducibly eliminated by spin-coating a thin
13 additional layer of PCBM on top of the BHJ, which improves the extraction of electrons
14 from the device by ‘repairing’ the poor contact at the PCBM-cathode interface.
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25 The solar cells we describe below were fabricated by first spin-coating a ≤ 50 -nm thick
26 layer of poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS, Clevios P
27 VP A14083) onto indium-doped tin oxide (ITO) coated glass substrates (TFD, Inc.). The ac-
28 tive layers were then deposited by spin-coating (at 600 RPM for 5 min) an *o*-dichlorobenzene
29 (ODCB) solution of P3HT (2% w/v, Rieke Metals)) and PCBM (Nano-C) mixed in a 1:0.8
30 weight ratio to produce an ~ 140 -nm thick BHJ film. We then thermally evaporated 70-nm
31 thick aluminum (99.99%, Kurt J. Lesker) cathodes onto the structures through a shadow
32 mask, producing devices with an active area of 4 mm². The devices discussed in this paper
33 were thermally annealed at 150 °C for 20 mins while covered with a shallow glass Petri dish.
34 Additional details of our methods for substrate preparation, device fabrication and device
35 characterization under AM-1.5 illumination are described in the Supplementary Informa-
36 tion. Atomic force microscopy (AFM) was carried out using a Nanoscope V Dimension 5000
37 (Veeco Digital Instruments) in ambient conditions. Antimony *n*-doped silicon cantilevers
38 (TESPW, Veeco Probes) with spring constants of 42 N/m, first longitudinal resonance fre-
39 quencies between 230-410 kHz, and nominal tip radii of 8 nm were employed in tapping
40 mode. Simultaneous height and phase images were acquired and reproduced across multiple
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55 To understand the nature of the charge extraction and carrier mobility in P3HT:PCBM
56 solar cells with the S-curve, we used photo-CELIV,[16, 19] a technique that has been pre-
57 viously applied to examine the transport and recombination of photogenerated carriers in
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3 polymer:fullerene thin-film BHJ solar cells.[20–23] In our photo-CELIV measurements, we
4 partly compensated for the built-in potential that exists within the device by first applying
5 a small, constant offset bias (+0.07 to +0.18 V, depending on the sample) to help prevent
6 photogenerated charge carriers from migrating toward their respective electrodes. We then
7 illuminated the sample with a ~ 7 -ns pulse of light at 500 nm from a 10-Hz N₂-pumped dye
8 laser to create charge carriers within the polymer-fullerene blend film. After an adjustable
9 time delay, t_{del} , controlled by a digital delay generator (SRS DG535), we applied a reverse-
10 bias voltage ramp with a slope of 4 V/100 μs created with an arbitrary function generator
11 (SRS DS345) to extract any carriers that had not recombined. We then monitored the
12 resulting current, j , as a function of time using the 50 Ω input of a digital oscilloscope (Tek-
13 tronix DPO3014). The top panel of Figure 2a depicts the time sequence of the applied light
14 pulse and voltages. The bottom panel of Figure 2a shows a typical corresponding measured
15 current transient. The shaded part of the current transient, with amplitude labelled Δj ,
16 represents the extraction of the photogenerated charge carriers. This signal appears on top
17 of the rectangular capacitive current, j_0 , which has a typical magnitude of 0.3 mA/cm² for
18 our devices; in the data presented below, we have subtracted the capacitive current, mea-
19 sured when the light pulse is blocked, to better focus on the dynamics of the photogenerated
20 carriers extracted by the voltage ramp.
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35 Figure 2b shows photo-CELIV current transients, collected at different values of t_{del} , for
36 a P3HT:PCBM BHJ solar cell that has an S-shaped J - V curve under AM-1.5 illumination,
37 similar to that shown by the dashed curve in Figure 1. The current transients clearly show
38 two distinct extraction peaks at $t_{\text{max}} \approx 2$ and 12 μs , a signature that to the best of our
39 knowledge has not been previously reported for polymer-based photovoltaic devices. The
40 fact that two peaks are observed in the current transient signifies that the extraction of
41 holes at the anode/BHJ interface occurs at a different rate than the extraction of electrons
42 at the BHJ/cathode interface. We can get a rough estimate the mobilities of the carriers
43 responsible for the two peaks in the current transient from
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$$\mu = 2d^2[3At_{\text{max}}^2(1 + \Delta j/j_0)]^{-1} \quad (1)$$

52 where μ is the charge carrier mobility, d is the thickness of the active layer, A is the slope
53 of the extraction voltage ramp, t_{max} is the time at which the extracted reaches its peak, Δj
54 is the current extraction peak height, and j_0 is the value of the dark, capacitive current, as
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3 shown in Fig. 2a.[24] For the data in Fig. 2b, we find that $\mu_{\text{fast}} = 2 \times 10^{-4} \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ and
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5 $\mu_{\text{slow}} = 1 \times 10^{-5} \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ for the fast and slow carriers, respectively. We note that the
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7 derivation of Eq. 1 assumes that one of the contacts is blocking and the other is Ohmic,
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9 which is clearly not a good assumption for the ambipolar carrier extraction we observe.
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11 Thus, the mobilities we extract using Eq. 1 are likely not highly accurate, but the order
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13 of magnitude difference in their values serves as a clear indicator that the main problem
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15 responsible for the S-curve is misbalanced extraction of the two carriers.

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17 To elucidate the reasons underlying this imbalanced charge extraction, we examined the
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19 structural characteristics of the BHJ/cathode interface. Figure 3a shows an AFM phase
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21 image of the surface of the active layer of the P3HT:PCBM BHJ solar cell whose photo-
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23 CELIV transients are shown in Fig. 2b. The image shows rice-like crystalline polymer
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25 domains with an average diameter of $11.6 \pm 2.7 \text{ nm}$ and the film has an rms surface roughness
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27 of 0.68 nm. These features are statistically identical to those obtained from the surface of
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29 pure P3HT films cast from this same solvent, as shown in Fig. 3b. It is striking that the
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31 surface of the S-curve device shows none of the features assigned by others[25] to PCBM-
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33 rich domains, which are visible on the surface of BHJ films that comprise the active layer
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35 of solar cells that do not exhibit the S-curve behavior. We show an AFM image of the top
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37 surface of a BHJ film that gives ‘typical’ device performance in Fig. 3c: the rice-like grain
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39 structure of the polymer is suppressed since PCBM breaks up the nanoscale crystallinity of
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41 the P3HT. The dark features in this phase image, which correspond to bumps in topography,
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43 indicate that PCBM-rich domains are present at the top surface of the film, consistent with
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45 transmission electron microscopy work by others.[26] The fact that the surface of BHJ films
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47 in good devices show evidence for PCBM at the top surface while BHJ films that show
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49 the S-curve have a top surface that is nearly entirely composed of P3HT explains the poor
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51 performance of these devices: a good P3HT:PCBM solar cell needs PCBM on the surface
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53 to be in contact with the cathode for efficient extraction of electrons from the device. In
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55 addition, placing excess PCBM at the cathode interface may prevent possible chemical
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57 reactions between P3HT and the cathode, which have been suggested by others to cause
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59 poor device performance.[11, 13]

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61 We believe that the lack of PCBM on the surface of BHJ films that give poor photovoltaic
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63 performance results from vertical phase segregation that takes place during spin-coating; this
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65 phase segregation drives PCBM towards the bottom of the film, away from the top surface

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3 in contact with the cathode.[17, 18] This type of behavior is expected for BHJ systems
4 in which the PCBM component is much more soluble in the ODCB solvent than in the
5 P3HT polymer.[27] During spin-coating, particularly under very dry conditions, the solvent
6 evaporates first from the top of the film but is retained towards the substrate. This causes
7 the PCBM component to segregate toward the solvent-rich (*i.e.*, substrate) side of the BHJ
8 film. In accordance with this idea, techniques such as solvent annealing[4, 28] or the addition
9 of slowly evaporating solvent additives[7] are known to modify the vertical distribution of
10 solvent (and hence PCBM) in the film during the evaporation process and thus can greatly
11 improve BHJ device performance:[29] the presence of solvent vapor in the atmosphere during
12 spin-coating can reduce the solvent gradient within the film during evaporation and thus
13 reduce the amount of vertical phase separation. Unfortunately, the details of the solvent
14 evaporation kinetics can be difficult to control experimentally, so that small changes in the
15 degree of vertical phase separation of the polymer and fullerene components in the BHJ
16 film lead to irreproducible device performance that in extreme cases results in the S-curve.
17 In the Supplementary Information, we show that changing the drying conditions following
18 spin-coating our P3HT:PCBM BHJ active layers can have a large effect on the S-curve: BHJ
19 samples that were placed under vacuum to remove solvent from the film immediately after
20 spin-coating are highly likely to show the S-curve, whereas samples prepared under identical
21 conditions that were allowed to dry in the presence of solvent vapor almost never show the
22 S-curve and have much more reproducible J - V characteristics.

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39 To further test our hypothesis that changes in the film's vertical composition profile
40 due to subtleties in the spin-coating process are what is responsible for the S-curve, we
41 investigated whether or not we could improve both device performance and reproducibility
42 by altering the film structure to increase the amount of PCBM at the top surface of the film
43 in contact with the cathode. To do this, we simply spin-coated additional PCBM from an
44 0.5% w/v solution in dichloromethane (DCM) on top of the active BHJ layer at 4000 rpm
45 for 10 seconds, producing a \sim 20-nm thick PCBM overlayer. We have shown elsewhere that
46 the morphology of P3HT films cast from ODCB is not altered when a DCM solution is spun
47 on top of the P3HT layer.[30] In Figure 3d, we show an AFM phase image of a BHJ film
48 onto which a PCBM overlayer has been spun from DCM; the surface shows all the same
49 features (occasional PCBM nanocrystals embedded in an amorphous PCBM background)
50 as the PCBM overlayers we spun onto pure P3HT films in our previous work.[30] Thus, the
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3 AFM image in Fig. 3d strongly suggests that the additional PCBM completely covers the
4 surface of the BHJ film. Based on this result and the device data shown below, we believe
5 that the act of spin-coating a PCBM overlayer onto a BHJ film accomplishes two things: it
6 introduces additional PCBM onto the top surface of the BHJ film, and it leads to dissolution
7 and redistribution of the PCBM within the BHJ, (partially) reversing some of the undesired
8 vertical phase segregation.
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14 The dot-dashed curve in Figure 1 shows the J - V characteristics of the solar cell fabricated
15 from the BHJ film with the PCBM overlayer whose AFM image is shown in Figure 3d; this
16 device was made at the same time and under identical conditions as the device without the
17 PCBM overlayer whose photo-CELIV transients are shown in Fig. 2b. The figure makes it
18 clear that the addition of the PCBM overlayer has ‘repaired’ the S-curve. We note that *all*
19 of the BHJ devices with PCBM overlayers that we have tested, including those with BHJ
20 thicknesses in the range of 80-115 nm, show high fill-factors: the addition of the PCBM
21 overlayer not only eliminates the S-curve, but it also creates devices whose performance
22 characteristics are much more reproducible than those fabricated with the BHJ layer alone.
23 We also found that spinning PCBM overlayers from solutions with lower concentrations
24 prevented the S-curve but did not increase device performance to the same extent as devices
25 with overlayers spun from 0.5% w/v solutions.
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36 We can see the reason that the PCBM overlayer improves device performance in Fig. 2c,
37 which shows photo-CELIV transients for the solar cell with the PCBM overlayer whose J - V
38 characteristics are shown as the dot-dashed curve in Fig. 1 and whose AFM image is shown
39 in Fig. 3d. Figure 2c shows only a single peak in the extracted current at early times,
40 $t_{\max} = 1.8 \mu\text{s}$, which via Eq. 1 corresponds to an effective carrier mobility of $\mu_{\text{avg}} = 5 \times 10^{-4}$
41 $\text{cm}^2\text{V}^{-1}\text{s}^{-1}$. The fact that we see only a single peak suggests that electrons and holes are
42 being extracted at roughly the same rate from this device. The higher calculated effective
43 mobility relative to the device in Fig. 2b indicates that we have improved the extraction
44 of both types of carriers: the improvement in electron extraction prevents space-charge
45 build-up, which in turn improves the extraction of holes. Thus, we assign the fast and slow
46 extraction peaks in Fig. 2b to the extraction of holes and electrons, respectively. Usually,
47 photo-CELIV measurements cannot identify carrier types, but in this case, we can make the
48 assignment because we know that electron extraction is specifically improved by inserting the
49 additional PCBM layer between the BHJ and cathode, causing the longer-time extraction
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3 peak to disappear. We note that the magnitude of the extracted charge in Fig. 2c is less
4 than that in Fig. 2b because the device with the PCBM overlayer has better contacts that
5 allow a higher fraction of the photogenerated charges to leak out of the active layer during
6 t_{del} , as discussed in more detail in the Supplementary Information.
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10 In summary, we have investigated the reasons that the performance of P3HT:PCBM
11 BHJ solar cells fabricated using nominally identical conditions is irreproducible, which in
12 the most extreme cases leads to poor devices that have an S-shaped J - V curve. Using
13 the transient current technique photo-CELIV, we found that mismatched hole and electron
14 charge-carrier extraction rates are the cause of the reduced fill-factor S-curve. AFM images
15 of the BHJ active layer in devices that exhibit the S-curve showed that the composition of
16 the top surface of the film is dominated by P3HT, the result of vertical phase segregation
17 that is controlled by subtle factors in the solvent evaporation kinetics during spin-coating.
18 When this type of vertical phase separation occurs, electron extraction is hindered by the
19 poor contact between the PCBM component of the BHJ and the cathode. We were able
20 to repair this poor contact by spin-coating an additional PCBM layer onto the BHJ film,
21 reversing some of the undesired vertical phase segregation and improving electron collection
22 at the cathode. Perhaps most importantly, unlike single-layer BHJ devices, the performance
23 of BHJ solar cells with a PCBM overlayer are much more reproducible from batch-to-batch
24 and day-to-day, something that is a key requirement if polymer-based photovoltaics are to
25 fulfill their promise in commercial applications.
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41 ACKNOWLEDGMENTS

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47 testing. Supplementary Information for this manuscript is available in the form of an addi-
48 tional PDF file that describes both the details of the experimental techniques used in this
49 work and the results of additional experiments that are mentioned briefly in the text. This
50 information is available free of charge via the Internet at <http://pubs.acs.org>.
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-FIGURE CAPTIONS-

FIG. 1: Current-voltage characteristics under AM-1.5 illumination for P3HT:PCBM BHJ solar cells. The solid curve shows the performance of a ‘typical’ ITO/PEDOT:PSS/P3HT:PCBM/Al BHJ solar cell with an ~ 140 -nm thick active layer, which exhibits a good fill-factor (50%) and reasonable power conversion efficiency (PCE). The dashed curve shows the J - V behavior of a device fabricated under nominally identical conditions as that for the solid curve, but which has a low fill-factor (25%) and PCE due to the presence of the ‘S-curve’; even when the S-curve is not present, the J - V curves of cells made on different days using the same nominal processing conditions are often different. The dot-dashed curve shows the J - V characteristics of a BHJ solar cell fabricated at the exact same time as the S-curve device whose J - V behavior is shown by the dashed curve, but which had an additional PCBM overlayer spun on top of the BHJ film, restoring the good fill-factor (54%) and PCE. In addition to the improvement in performance, we have never fabricated a BHJ device with a PCBM overlayer that shows the S-curve, and we find that devices with PCBM overlayers give much more reproducible J - V behavior.

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FIG. 2: a) Upper Panel: Timing schematic for the application of the light pulse and voltage ramp in photo-CELIV measurements. Lower Panel: Typical photo-CELIV current transient showing both capacitive (j_0) and photoextracted current (shaded area with amplitude Δj). b) Photo-CELIV current transients, $j - j_0$, collected at different values of t_{del} for the P3HT:PCBM BHJ device whose J - V characteristics are given by the dashed curve in Fig. 1. The two peaks indicated by the arrows correspond to the extraction of fast and slow carriers; we assign the fast peak to hole extraction and the slow peak to electron extraction for the reasons outlined in the text. c) Photo-CELIV current transients, $j - j_0$, collected at different values of t_{del} for a P3HT:PCBM device prepared at the same time as that in panel b except that a ~ 20 -nm thick PCBM overlayer was spin-cast on top of the BHJ film before deposition of the cathode; the J - V curve of this device is shown by the dot-dashed curve in Fig. 1. The presence of a single peak indicates that electrons and holes are being extracted at roughly the same rate. The larger magnitude of the current transients and slower decay rate of the extracted current in panel b relative to panel c is a direct result of the improved electron extraction for the device in panel c, which causes more carriers to leak out of the active layer for a given value of t_{del} . (See the Supplementary Information for additional details)

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FIG. 3: AFM tapping-mode phase images of the top surface of a) the P3HT:PCBM BHJ film whose S-curve J - V behavior is shown by the dashed curve in Fig. 1 and whose photo-CELIV transients are shown in Fig. 2b. The surface roughness of this film is 1.55 nm, and the diameter of the P3HT nanocrystalline domains is 11.6 ± 2.7 nm; b) a pure P3HT film spun from ODCB, whose surface topography and phase image features are statistically identical to the BHJ film shown in panel a, indicating that the surface of the S-curve BHJ film is nearly entirely composed of P3HT; c) a P3HT:PCBM film whose J - V characteristics show the ‘typical’ behavior similar to the solid curve in Fig. 1. The ‘rice-grain’ like structure seen in panels a and b is absent because the presence of PCBM near the top surface of the film interferes with P3HT’s ability to form nanocrystalline domains. The large dark features (which correspond to bumps in the topography that give an rms surface roughness of 1.5 nm) correspond to PCBM-rich regions at the surface of the film. d) the P3HT:PCBM BHJ film with the ~ 20 -nm PCBM overlayer spin-cast from DCM whose photo-CELIV transients are shown in Fig. 2c and whose J - V behavior is shown as the dot-dashed curve in Fig. 1. The image shows a few 6.6-nm average diameter nanocrystallites of PCBM on an amorphous PCBM background, indicating that the mostly amorphous PCBM overlayer entirely covers the surface of the underlying BHJ.

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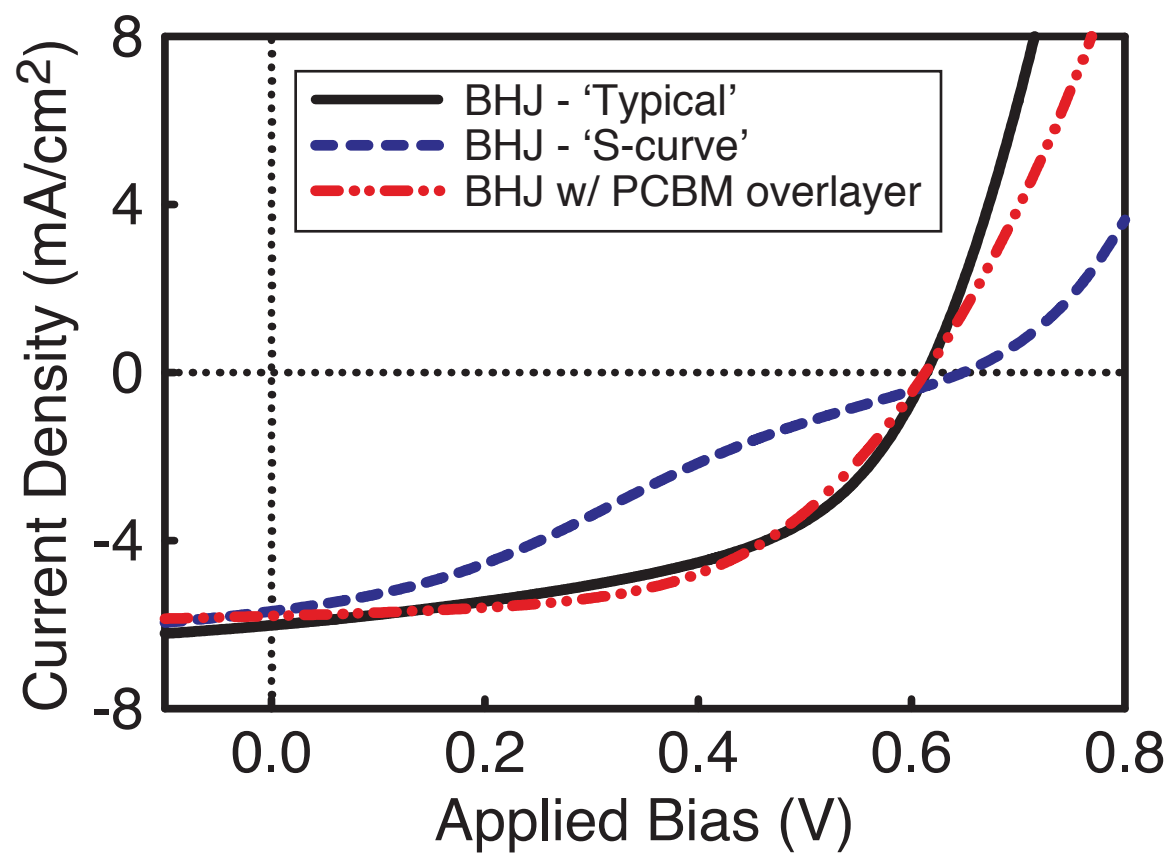
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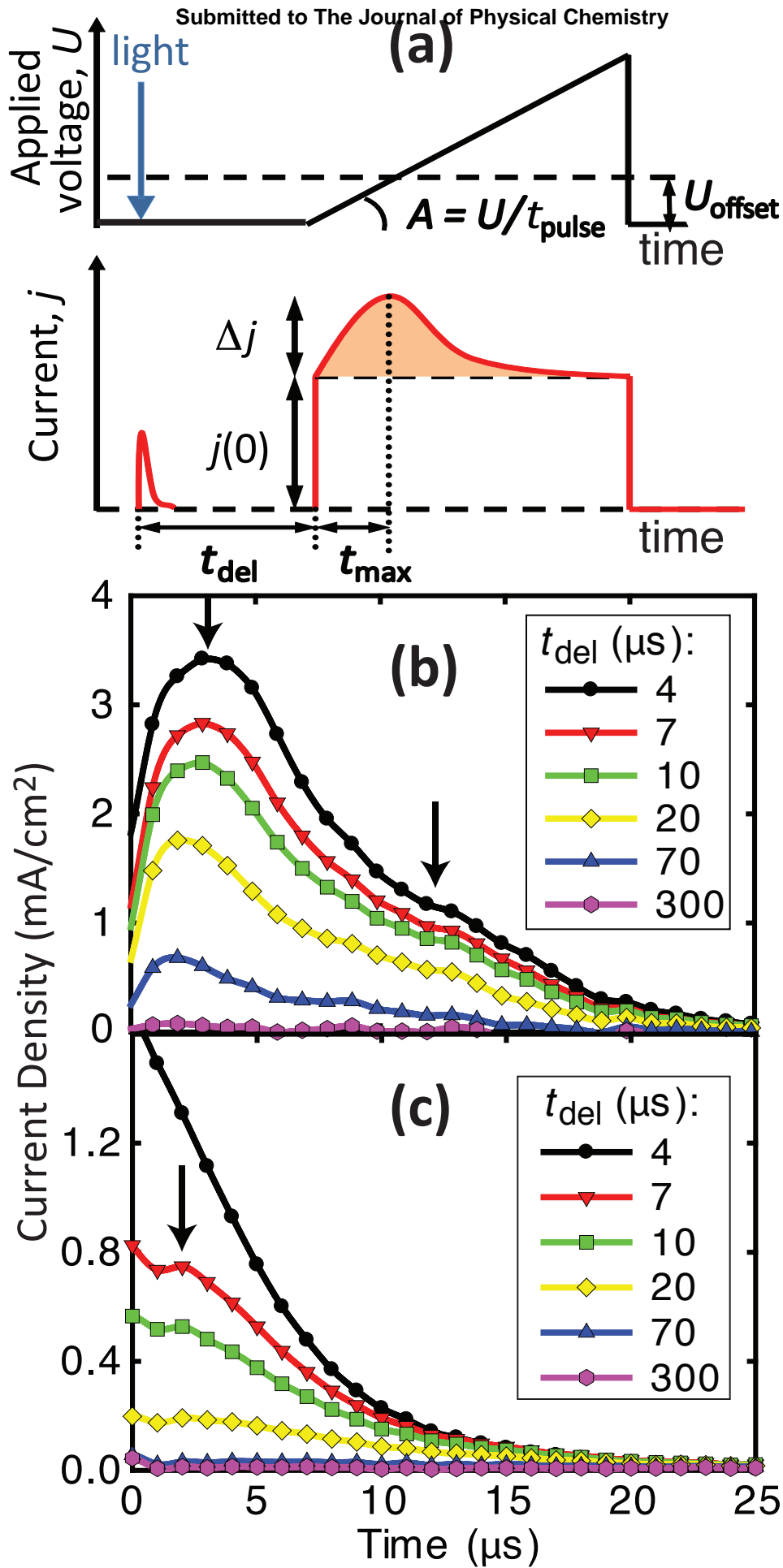
- 9
10 [1] Mayer, A. C.; Scully, S. R.; Hardin, B. E.; Rowell, M. W.; McGehee, M. D. *Materials Today*
11 **2007**, *10*, 28–33
12
13 [2] Gunes, S.; Neugebauer, H.; Sariciftci, N. S. *Chem. Rev.* **2007**, *107*, 1324–1338
14
15 [3] Park, S. H.; Roy, A.; Beaupre, S.; Cho, S.; Coates, N.; Moon, J. S.; Moses, D.; Leclerc, M.;
16 Lee, K.; Heeger, A. J. *Nat. Photon.* **2009**, *3*, 297–302
17
18 [4] Chen, L.-M.; Hong, Z.; Li, G.; Yang, Y. *Adv. Mater.* **2009**, *21*, 1434–1449
19
20 [5] Hoppe, H.; Sariciftci, N. S. *J. Mater. Chem.* **2006**, *16*, 45–61
21
22 [6] Reyes-Reyes, M.; Kim, K.; Carroll, D. L. *Appl. Phys. Lett.* **2005**, *87*, 083506–3
23
24 [7] Peet, J.; Kim, J. Y.; Coates, N. E.; Ma, W. L.; Moses, D.; Heeger, A. J.; Bazan, G. C. *Nat.*
25 *Mater.* **2007**, *6*, 497–500
26
27 [8] Lee, J. K.; Ma, W. L.; Brabec, C. J.; Yuen, J.; Moon, J. S.; Kim, J. Y.; Lee, K.; Bazan, G. C.;
28 Heeger, A. J. *J. Am. Chem. Soc.* **2008**, *130*, 3619–3623
29
30 [9] Ma, W. L.; Yang, C. Y.; Gong, X.; Lee, K.; Heeger, A. J. *Adv. Funct. Mater.* **2005**, *15*,
31 1617–1622
32
33 [10] Reyes-Reyes, M.; Kim, K.; Dewald, J.; Lopez-Sandoval, R.; Avadhanula, A.; Curran, S.;
34 Carroll, D. L. *Org. Lett.* **2005**, *7*, 5749–5752
35
36 [11] Gupta, D.; Bag, M.; Narayan, K. S. *Appl. Phys. Lett.* **2008**, *92*, 093301–3
37
38 [12] Bisquert, J.; Garcia-Belmonte, G.; Munar, A.; Sessolo, M.; Soriano, A.; Bolink, H. J. *Chem.*
39 *Phys. Lett.* **2008**, *465*, 57–62
40
41 [13] Glatthaar, M.; Riede, M.; Keegan, N.; Sylvester-Hvid, K.; Zimmermann, B.; Niggemann, M.;
42 Hirsch, A.; Gombert, A. *Sol. Energy Mater. Sol. Cells* **2007**, *91*, 390–393
43
44 [14] Glatthaar, M.; Mingirulli, N.; Zimmermann, B.; Ziegler, T.; Kern, R.; Niggemann, M.; Hin-
45 sch, A.; Gombert, A. *Phys. Stat. Sol. A* **2005**, *202*, R125–R127
46
47 [15] Jin, H.; Tuomikoski, M.; Hiltunen, J.; Kopola, P.; Maaninen, A.; Pino, F. *J. Phys. Chem. C*
48 **2009**, *113*, 16807–16810
49
50 [16] Juska, G.; Arlauskas, K.; Viliunas, M.; Genevicius, K.; Osterbacka, R.; Stubb, H. *Phys. Rev.*
51 *B* **2000**, *62*, R16235
52
53
54
55
56
57
58
59
60

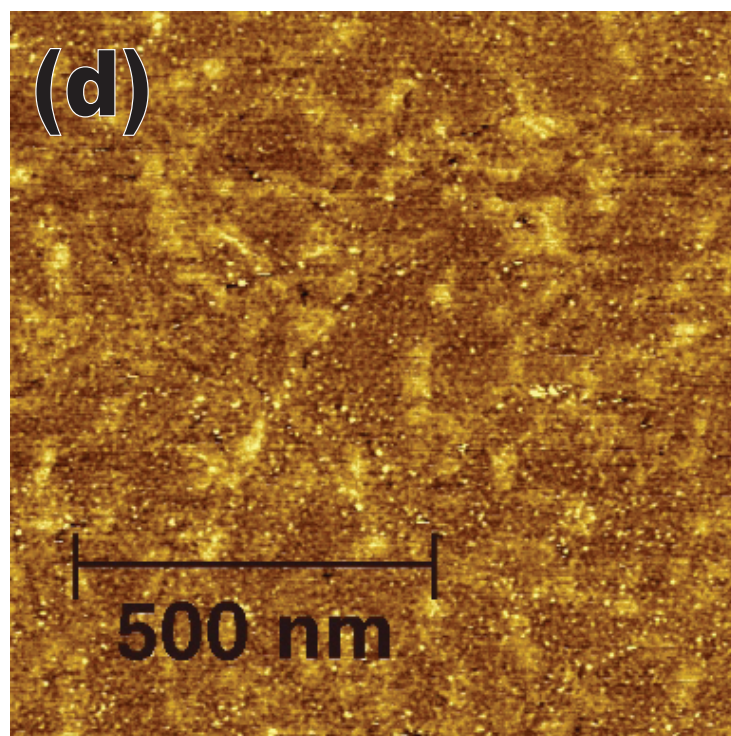
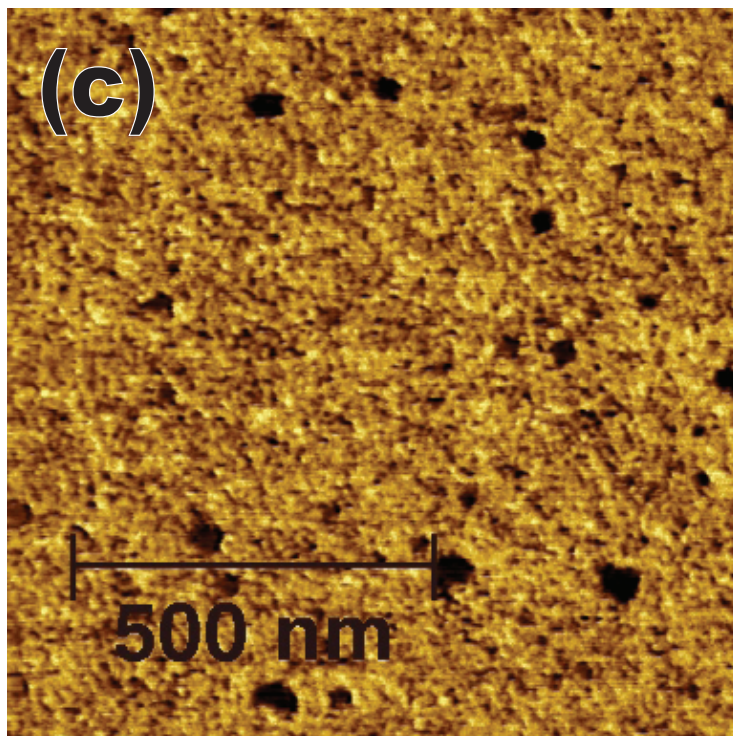
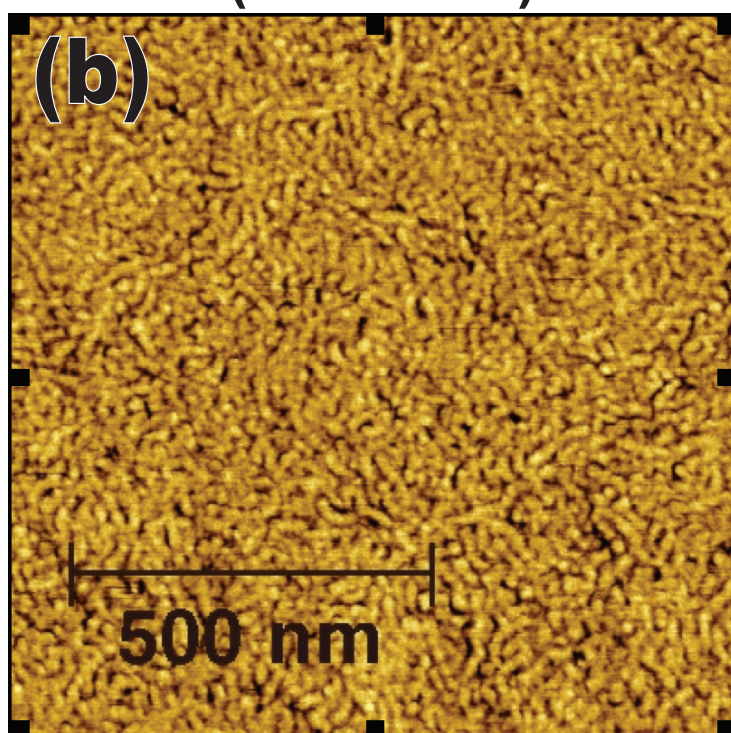
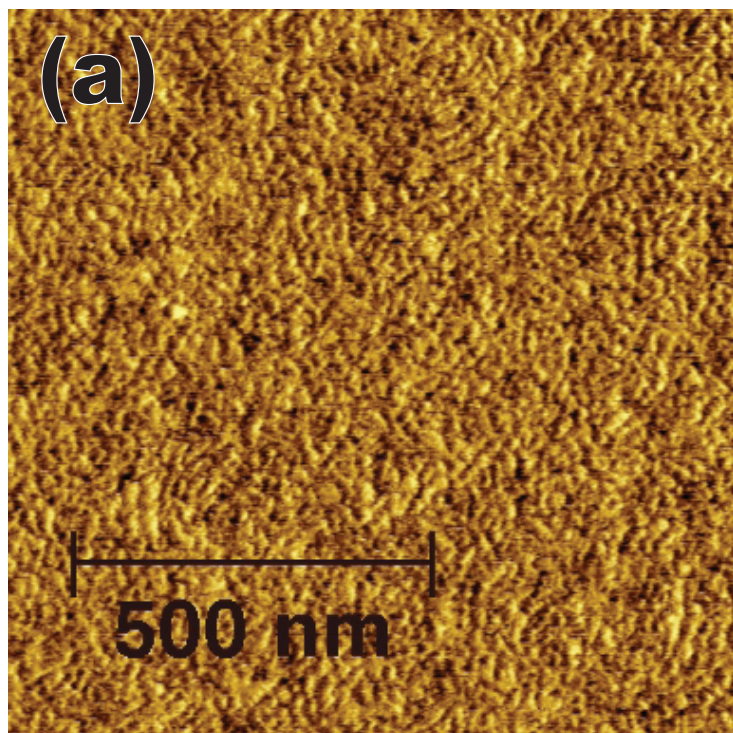
- 1
2
3 [17] Campoy-Quiles, M.; Ferenczi, T.; Agostinelli, T.; Etchegoin, P. G.; Kim, Y.; Anthopoulos, T. D.; Stavrinou, P. N.; Bradley, D. D. C.; Nelson, J. *Nat. Mater.* **2008**, *7*, 158–164
- 4
5
6
7 [18] Xu, Z.; Chen, L.-M.; Yang, G.; Huang, C.-H.; Hou, J.; Wu, Y.; Li, G.; Hsu, C.-S.; Yang, Y.
8
9 *Adv. Funct. Mater.* **2009**, *19*, 1227–1234
- 10
11 [19] Juska, G.; Arlauskas, K.; Viliunas, M.; Kocka, J. *Phys. Rev. Lett.* **2000**, *84*, 4946–4949
- 12
13 [20] Mozer, A. J.; Dennler, G.; Sariciftci, N. S.; Westerling, M.; Pivrikas, A.; Osterbacka, R.;
14
15 Juska, G. *Phys. Rev. B* **2005**, *72*, 035217
- 16
17 [21] Mozer, A. J.; Sariciftci, N. S.; Pivrikas, A.; Osterbacka, R.; Juska, G.; Brassat, L.; Bassler, H.
18
19 *Phys. Rev. B* **2005**, *71*, 035214
- 20
21 [22] Dennler, G.; Mozer, A. J.; Juska, G.; Pivrikas, A.; Osterbacka, R.; Fuchsbaauer, A.; Sari-
22
23 ciftci, N. S. *Org. Electron.* **2006**, *7*, 229–234
- 24
25 [23] Pivrikas, A.; Sariciftci, N. S.; Juska, G.; Osterbacka, R. *Prog. Photovoltaics* **2007**, *15*, 677–696
- 26
27 [24] Juska, G.; Viliunas, M.; Arlauskas, K.; Nekrasas, N.; Wyrsh, N.; Feitknecht, L. *J. Appl.*
28
29 *Phys.* **2001**, *89*, 4971–4974
- 30
31 [25] Li, G.; Shrotriya, V.; Yao, Y.; Huang, J.; Yang, Y. *J. Mater. Chem.* **2007**, *17*, 3126–3140
- 32
33 [26] Yang, X.; Loos, J.; Veenstra, S. C.; Verhees, W. J. H.; Wienk, M. M.; Kroon, J. M.; Michels, M.
34
35 A. J.; Janssen, R. A. J. *Nano Lett.* **2005**, *5*, 579–583
- 36
37 [27] Troshin, P. A.; Hoppe, H.; Renz, J.; Egginger, M.; Mayorova, J. Y.; Goryachev, A. E.; Peregu-
38
39 dov, A. S.; Lyubovskaya, R. N.; Gobsch, G.; Sariciftci, N. S.; Razumov, V. F. *Adv. Funct. Mater.*
40
41 **2009**, *19*, 779–788
- 42
43 [28] Li, G.; Yao, Y.; Yang, H.; Shrotriya, V.; Yang, G.; Yang, Y. *Adv. Funct. Mater.* **2007**, *17*,
44
45 1636–1644
- 46
47 [29] Dante, M.; Garcia, A.; Nguyen, T.-Q. *J. Phys. Chem. C* **2009**, *113*, 1596–1600
- 48
49 [30] Ayzner, A. L.; Tassone, C. J.; Tolbert, S. H.; Schwartz, B. J. *J. Phys. Chem. C* **2009**, *in*
50
51 *press*, jp-2009-050897

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**P3HT:PCBM
BHJ with 'S-curve'****Pure P3HT
(no PCBM)****P3HT:PCBM BHJ
without 'S-curve'****P3HT:PCBM BHJ
with PCBM overlayer**

