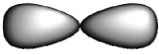

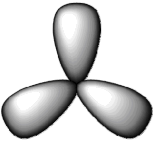
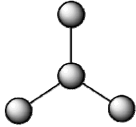
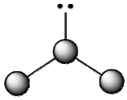
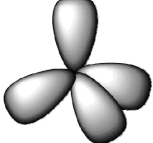
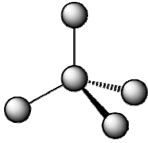
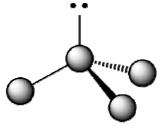
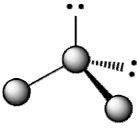
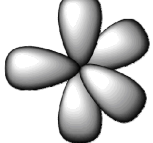
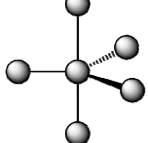
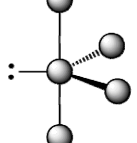
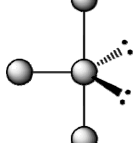
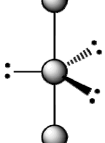
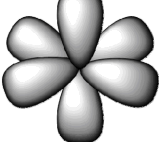
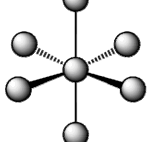
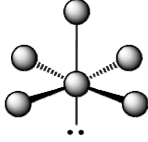
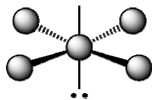


Hybridization (Central Atom)	Common Molecular Shapes			
				
sp	Linear			
				
sp ²	Trigonal Planar	Bent (~120°)		
				
sp ³	Tetrahedral	Trigonal Pyramidal	Bent (~109.5)	
				
sp ³ d	Trigonal Bipyramidal	See-Saw	T-Shaped	Linear
				
sp ³ d ²	Octahedral	Square Pyramidal	Square Planar	