

Last Name	<b>ANSWER</b>	First Name	<b>KEY</b>	MI
Student ID Number:				Total Score <b>35</b> / 30
Circle the name of your TA: Cari / Phil / Adam / Heather				
Discussion Section – Day:		Time:		

Chem 30A Fall 2004

**QUIZ #2 (BLUE)**  
(15 Min)

Weds Nov 10th

*INTERPRETATION OF THE QUESTIONS IS PART OF THE EXAM – DO NOT ASK FOR THE QUESTIONS TO BE EXPLAINED TO YOU*

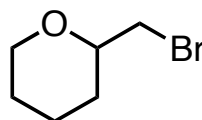
*USE CAPITAL LETTERS WHEN FILLING IN THE BOXES AND BE CLEAR – IF WE CAN'T FIGURE OUT WHAT A LETTER IS, IT WILL AUTOMATICALLY BE GRADED AS INCORRECT*

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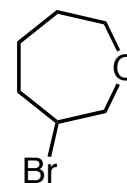
Q	1	2	3	4	5	6	7	8	9	10	Total
<b>X</b>	D	E	B	C	C	E	B	B	E	E	30

**ANSWER TO BONUS QUESTION**

This one more likely  $\Rightarrow$



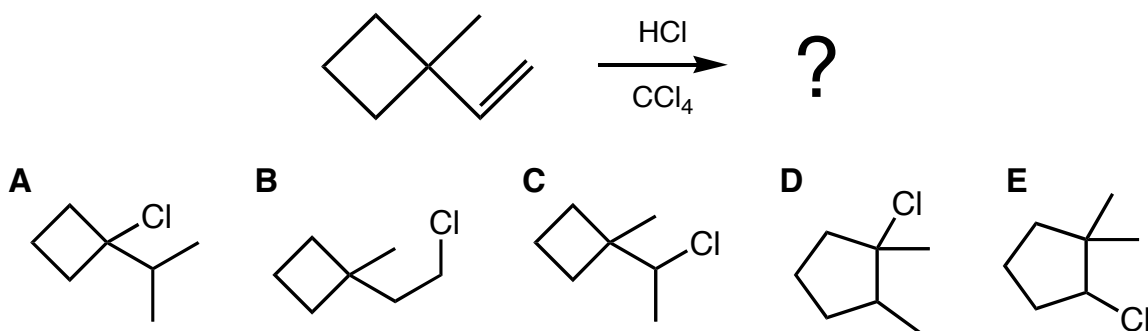
or



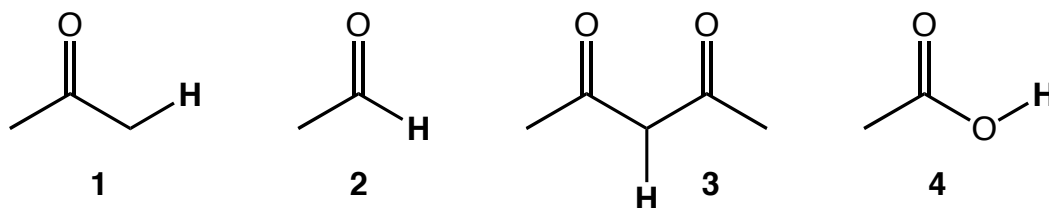
5

Questions 1–10 are worth 3 points each. The bonus is worth 5 points.

1. What is the MAJOR product of the reaction shown below?

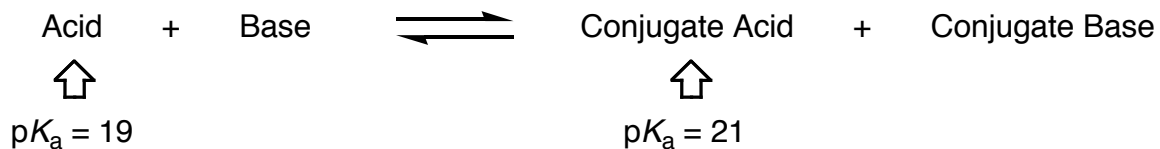


2. What is the order of acidity (from lowest  $pK_a$  value to highest  $pK_a$  value) of the bold hydrogen (H) atoms shown highlighted in the compounds drawn below?



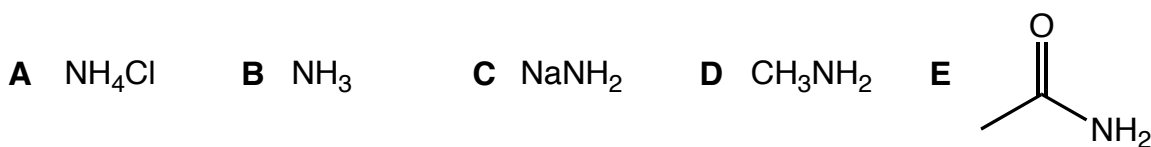
A 3, 4, 2, 1    B 4, 2, 1, 3    C 3, 1, 4, 2    D 2, 4, 3, 1    E 4, 3, 1, 2

3. The equilibrium constant ( $K_{eq}$ ) for the reaction shown below is...?

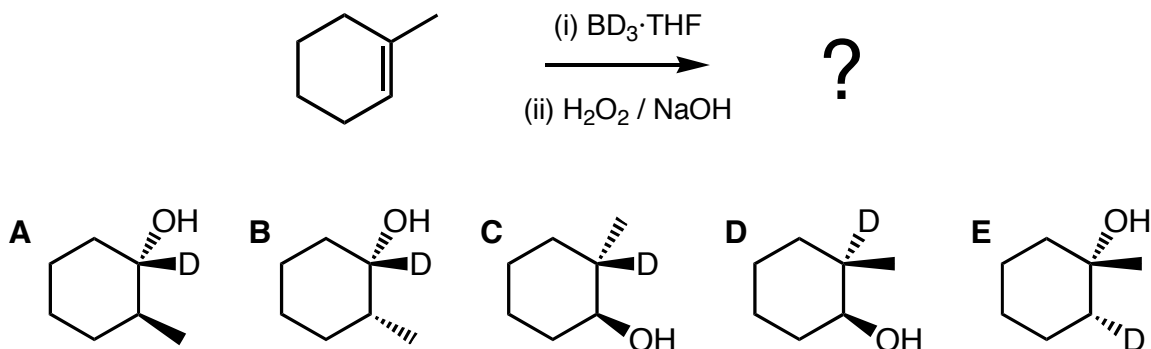


A 2    B 100    C -2    D 0.01    E 40

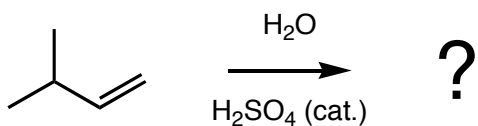
4. Which of the compounds drawn below is the strongest base?



5. Hydroboration/oxidation of 1-methylcyclohexene with deuterated borane·THF (reacts just like  $\text{BH}_3\cdot\text{THF}$ , but has D atoms instead of H) gives which compound?

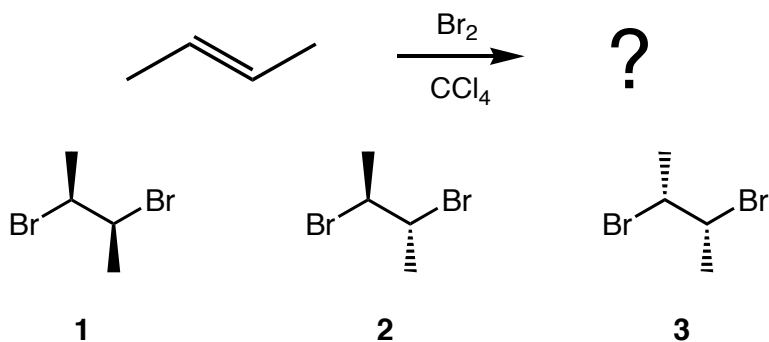


6. What is the MAJOR product of the reaction shown below?



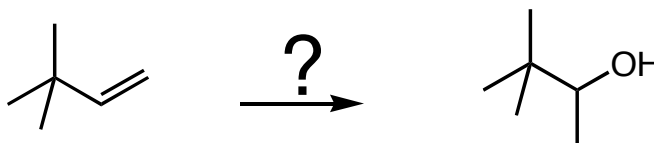
- A** 2,2-dimethyl-propan-1-ol  
**B** 3-methyl-butan-2-ol  
**C** 3-methyl-butan-1-ol  
**D** 2-methyl-butan-1-ol  
**E** 2-methyl-butan-2-ol

7. The product(s) of the reaction shown below are?



- A** Only 1    **B** Only 2    **C** Only 3    **D** Only 1 and 3    **E** 1, 2, and 3

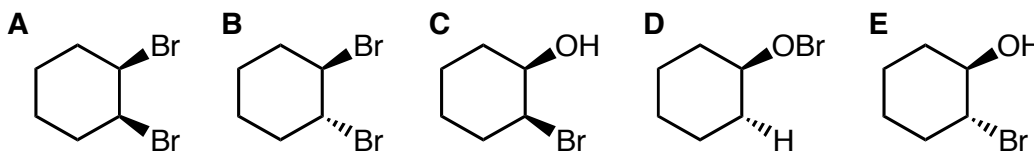
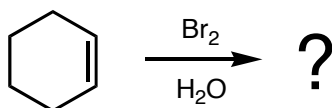
8. Which combination(s) of reagents would give rise to the reaction shown below?



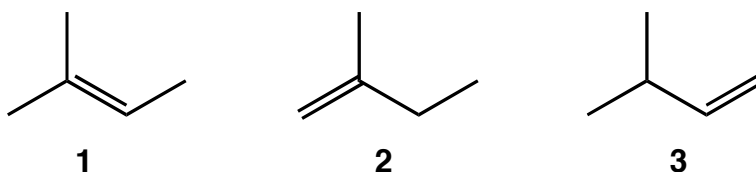
#1 (i)  $\text{BH}_3 \cdot \text{THF}$  (ii)  $\text{H}_2\text{O}_2 / \text{NaOH}$     #2 (i)  $\text{Hg}(\text{OAc})_2 / \text{H}_2\text{O}$  (ii)  $\text{NaBH}_4$     #3  $\text{H}_2\text{O} / \text{H}_2\text{SO}_4(\text{cat.})$

A Only #1    B Only #2    C Only #3    D Only #2 and #3    E #1, #2, and #3

9. What is the MAJOR product of the reaction shown below?

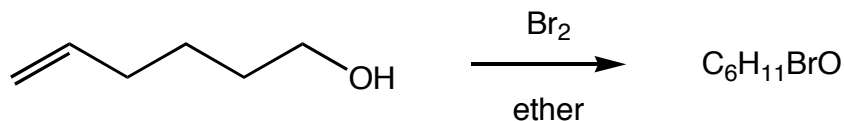


10. Which of the three alkenes drawn below will react with  $\text{HBr}$  in ether (solvent) to give 2-bromo-2-methylbutane?



A Only 1    B Only 2    C Only 3    D Only 1 and 3    E 1, 2, and 3

**BONUS:** The reaction of 5-hexen-1-ol with bromine in ether (solvent) gives a compound with the molecular formula  $\text{C}_6\text{H}_{11}\text{BrO}$ . Draw what you think the product of this reaction is ON THE FRONT COVER of this quiz in the box provided.



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Chem 30A Fall 2004

**QUIZ #2 (PINK)**  
(15 Min)

Weds Nov 10th

*INTERPRETATION OF THE QUESTIONS IS PART OF THE EXAM – DO NOT ASK FOR THE QUESTIONS TO BE EXPLAINED TO YOU*

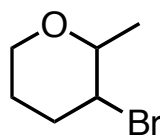
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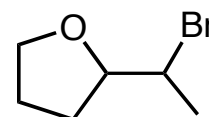
Q	1	2	3	4	5	6	7	8	9	10	Total
<b>X</b>	<b>C</b>	<b>B</b>	<b>E</b>	<b>A</b>	<b>E</b>	<b>D</b>	<b>C</b>	<b>D</b>	<b>B</b>	<b>A</b>	<b>30</b>

**ANSWER TO BONUS QUESTION**

6-membered ring more stable



or

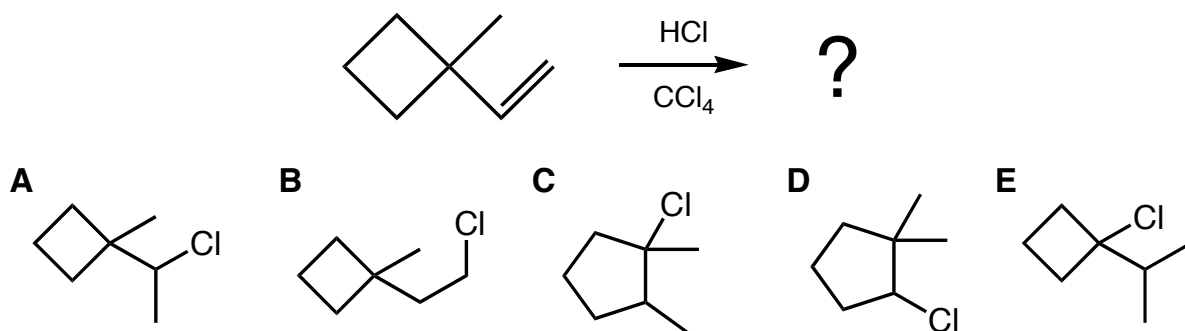


**5**

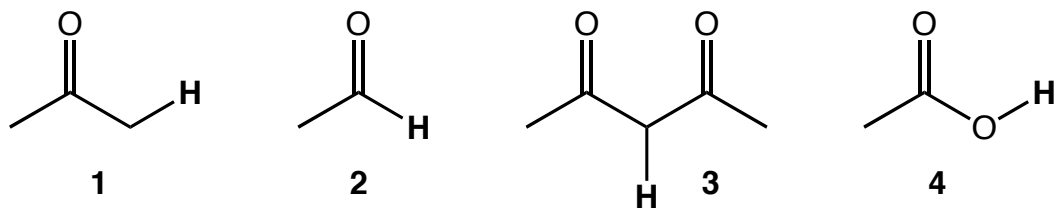
Being eaten by a crocodile is just like going to sleep... in a giant blender – Homer J Simpson

Questions 1–10 are worth 3 points each. The bonus is worth 5 points.

1. What is the MAJOR product of the reaction shown below?

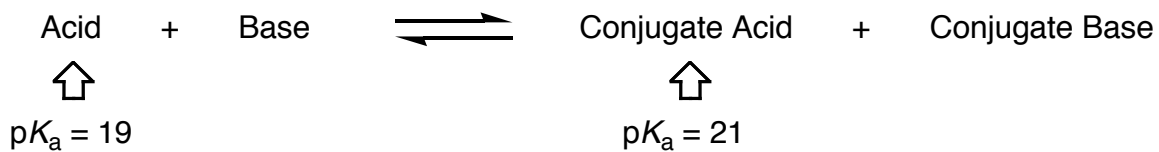


2. What is the order of acidity (from lowest  $pK_a$  value to highest  $pK_a$  value) of the bold hydrogen (H) atoms shown highlighted in the compounds drawn below?



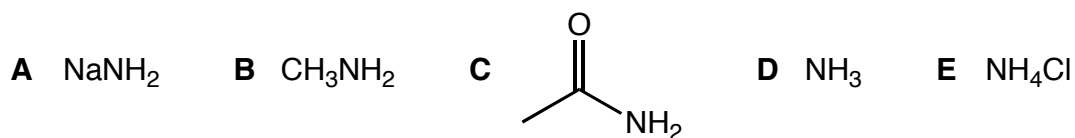
**A** 2, 4, 3, 1    **B** 4, 3, 1, 2    **C** 3, 1, 4, 2    **D** 3, 4, 2, 1    **E** 4, 2, 1, 3

3. The equilibrium constant ( $K_{eq}$ ) for the reaction shown below is...?

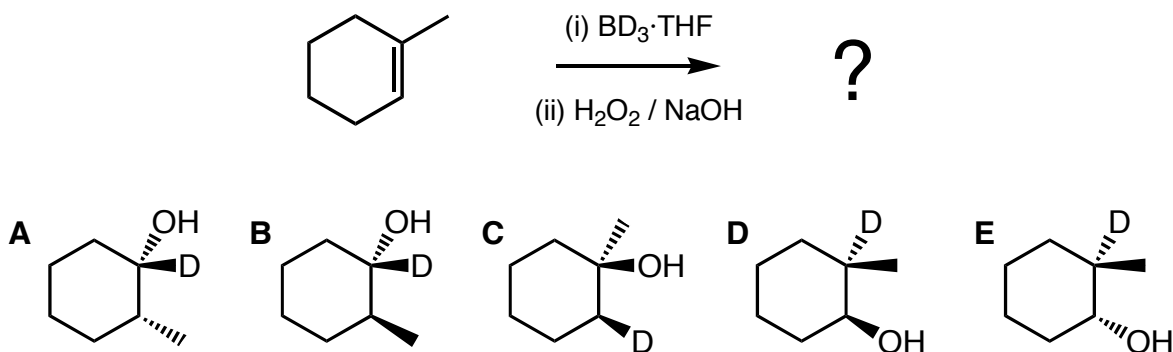


**A** -2    **B** 0.01    **C** 2    **D** 40    **E** 100

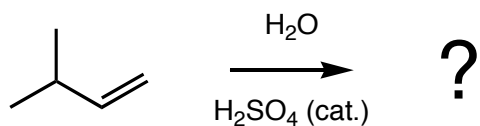
4. Which of the compounds drawn below is the strongest base?



5. Hydroboration/oxidation of 1-methylcyclohexene with deuterated borane·THF (reacts just like  $\text{BH}_3\cdot\text{THF}$ , but has D atoms instead of H) gives which compound?

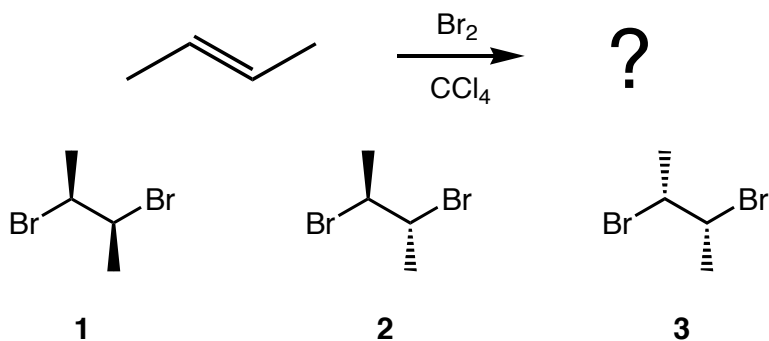


6. What is the MAJOR product of the reaction shown below?



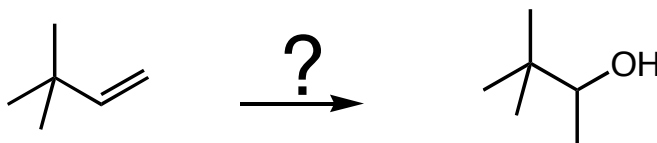
- A 2,2-dimethyl-propan-1-ol  
 B 3-methyl-butan-2-ol  
 C 3-methyl-butan-1-ol  
 D 2-methyl-butan-2-ol  
 E 2-methyl-butan-1-ol

7. The product(s) of the reaction shown below are?



- A Only 1 and 3    B 1, 2, and 3    C Only 2    D Only 1    E Only 3

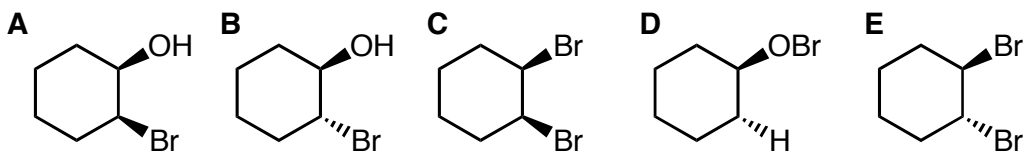
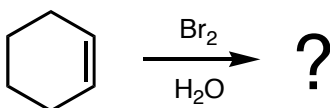
8. Which combination(s) of reagents would give rise to the reaction shown below?



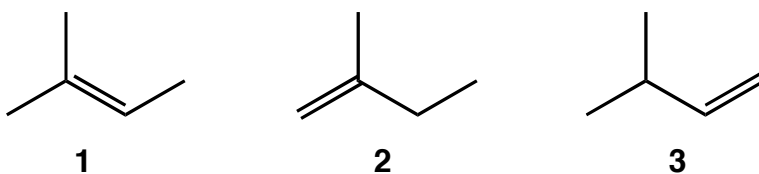
#1 (i)  $\text{BH}_3 \cdot \text{THF}$  (ii)  $\text{H}_2\text{O}_2 / \text{NaOH}$     #2 (i)  $\text{Hg}(\text{OAc})_2 / \text{H}_2\text{O}$  (ii)  $\text{NaBH}_4$     #3  $\text{H}_2\text{O} / \text{H}_2\text{SO}_4(\text{cat.})$

A #1, #2, and #3    B Only #2 and #3    C Only #1    D Only #2    E Only #3

9. What is the MAJOR product of the reaction shown below?

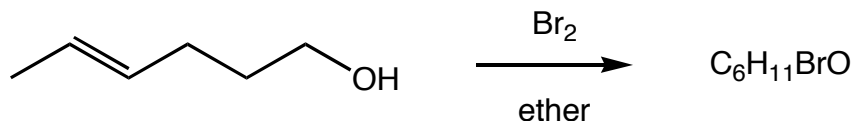


10. Which of the three alkenes drawn below will react with  $\text{HBr}$  in ether (solvent) to give 2-bromo-2-methylbutane?



A 1, 2, and 3    B Only 1 and 3    C Only 3    D Only 2    E Only 1

**BONUS:** The reaction of 4-hexen-1-ol with bromine in ether (solvent) gives a compound with the molecular formula  $\text{C}_6\text{H}_{11}\text{BrO}$ . Draw what you think the product of this reaction is ON THE FRONT COVER of this quiz in the box provided.





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Chem 30A Fall 2004

**QUIZ #2 (BUFF)**  
(15 Min)

Weds Nov 10th

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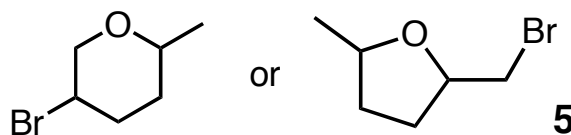
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Q	1	2	3	4	5	6	7	8	9	10	Total
<b>X</b>	<b>B</b>	<b>C</b>	<b>D</b>	<b>B</b>	<b>B</b>	<b>B</b>	<b>D</b>	<b>C</b>	<b>D</b>	<b>D</b>	<b>30</b>

**ANSWER TO BONUS QUESTION**

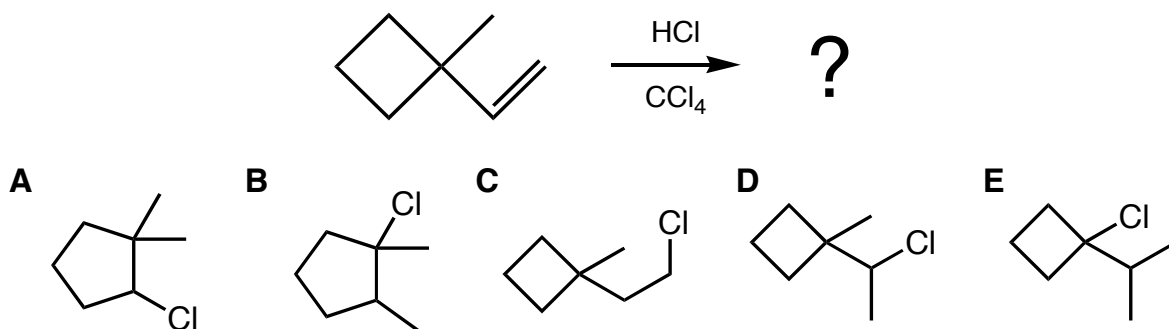
6-membered ring more stable, but 5-membered product puts  $\delta+$  on carbon better able to stabilize it in the transition state ( $2^\circ$  vs  $1^\circ$ )



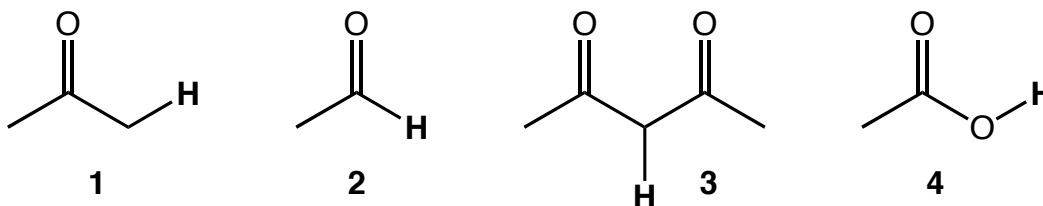
**5**

Questions 1–10 are worth 3 points each. The bonus is worth 5 points.

1. What is the MAJOR product of the reaction shown below?

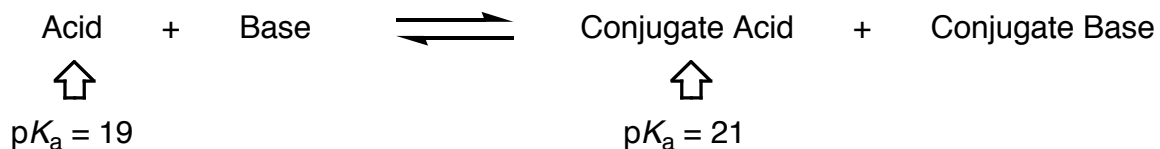


2. What is the order of acidity (from lowest  $pK_a$  value to highest  $pK_a$  value) of the bold hydrogen (H) atoms shown highlighted in the compounds drawn below?



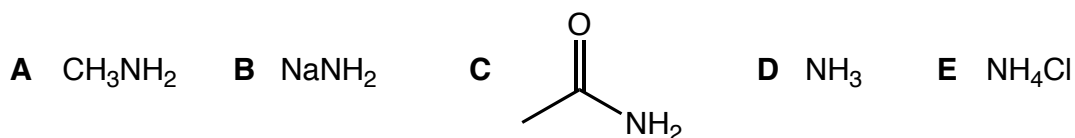
**A** 2, 4, 3, 1      **B** 3, 1, 4, 2      **C** 4, 3, 1, 2      **D** 3, 4, 2, 1      **E** 4, 2, 1, 3

3. The equilibrium constant ( $K_{eq}$ ) for the reaction shown below is...?

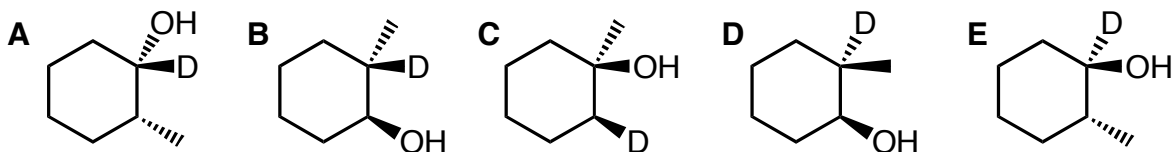
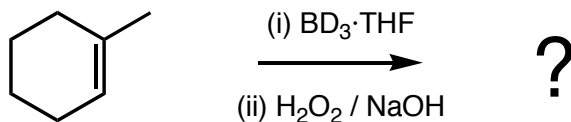


**A** 2      **B** -2      **C** 0.01      **D** 100      **E** 40

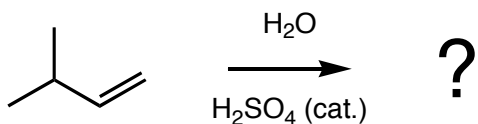
4. Which of the compounds drawn below is the strongest base?



5. Hydroboration/oxidation of 1-methylcyclohexene with deuterated borane·THF (reacts just like  $\text{BH}_3\cdot\text{THF}$ , but has D atoms instead of H) gives which compound?

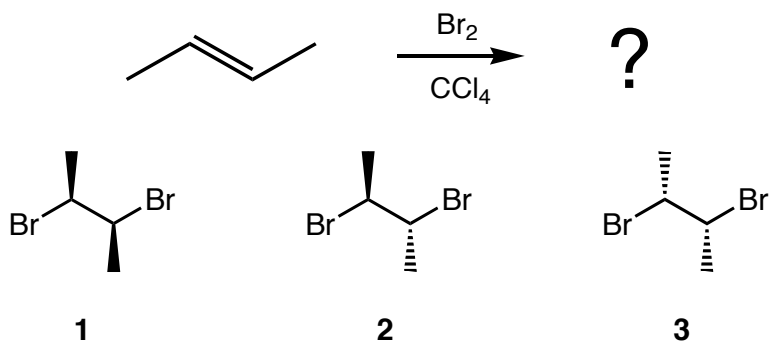


6. What is the MAJOR product of the reaction shown below?



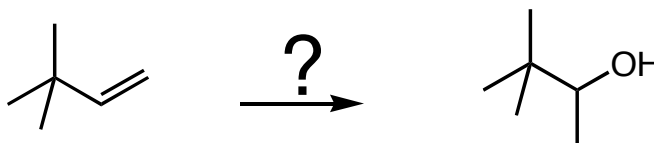
- A** 2,2-dimethyl-propan-1-ol  
**B** 2-methyl-butan-2-ol  
**C** 3-methyl-butan-2-ol  
**D** 3-methyl-butan-1-ol  
**E** 2-methyl-butan-1-ol

7. The product(s) of the reaction shown below are?



- A** Only 1 and 3    **B** 1, 2, and 3    **C** Only 3    **D** Only 2    **E** Only 1

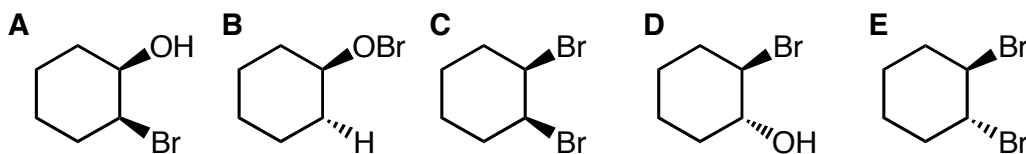
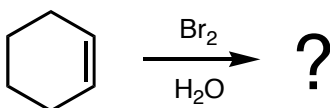
8. Which combination(s) of reagents would give rise to the reaction shown below?



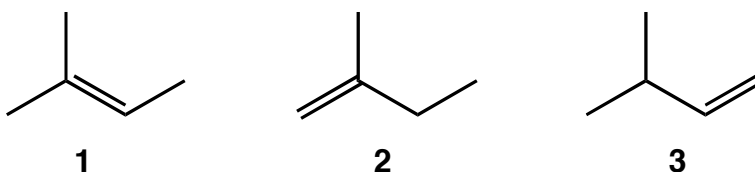
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A #1, #2, and #3    B Only #1    C Only #2    D Only #3    E Only #1 and #3

9. What is the MAJOR product of the reaction shown below?



10. Which of the three alkenes drawn below will react with  $\text{HBr}$  in ether (solvent) to give 2-bromo-2-methylbutane?



A Only 2    B Only 1 and 3    C Only 1    D 1, 2, and 3    E Only 3

**BONUS:** The reaction of 5-hexen-2-ol with bromine in ether (solvent) gives a compound with the molecular formula  $\text{C}_6\text{H}_{11}\text{BrO}$ . Draw what you think the product of this reaction is ON THE FRONT COVER of this quiz in the box provided.

