Last Name	First Name	MI
Student ID Number:		Total Score
Circle the name of your TA: HEATHER / KAUSHIK / CARI / RYAN		
Discussion Section – Day:	Time:	/ 25

Chem 30A Fall 2005

MIDTERM #2 SUPPLEMENT

(15 Min)

## Wed Nov 30th

INTERPRETATION OF THE QUESTIONS IS PART OF THE EXAM –
DO NOT ASK FOR THE QUESTIONS TO BE EXPLAINED TO YOU

ONLY ANSWERS WRITTEN IN THE BOXES PROVIDED WILL BE GRADED

\*\*\*DO NOT OPEN THIS EXAM UNTIL INSTRUCTED TO DO SO\*\*\*

Total / 25

"We didn't lose the game; we just ran out of time."

- Vince Lombardi

"All truth passes through three stages. First, it is ridiculed. Second, it is violently opposed.

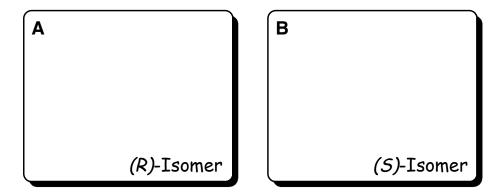
Third, it is accepted as being self-evident."

- Arthur Schopenhauer

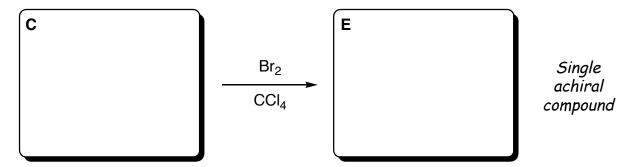
**Q1**. Four different cycloalkenes (**A–D**), each with the molecular formula  $C_5H_8$ , will yield methyl-cyclobutane when subjected to catalytic hydrogenation ( $H_2/Pt$  catalyst) – as shown below.

A, B, C, and D
$$\begin{array}{c}
H_2 \text{ (3 atm)} \\
\hline
Pt / C \text{ (cat.)}
\end{array}$$

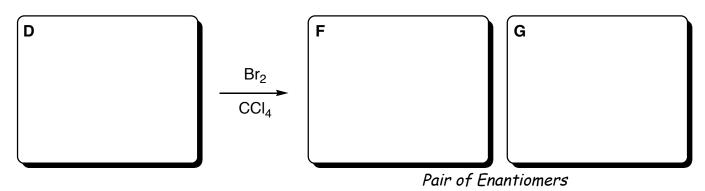
(a) Two of these cycloalkenes constitute a pair of enantiomers; the absolute configuration of the stereogenic center in compound  $\mathbf{A}$  is (R) and in  $\mathbf{B}$  it is (S). Draw these compounds below. (6 pt)



- (b) When cycloalkenes C & D are reacted with Br<sub>2</sub> in CCl<sub>4</sub>, different results are observed:
- (i) Cycloalkene C reacts to form a single achiral product (E) draw these compounds in the appropriate boxes below: (4 pt)

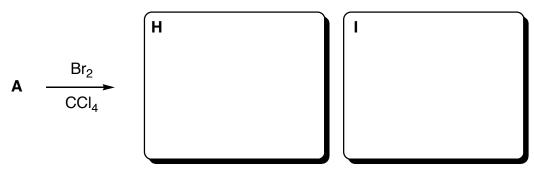


(ii) Cycloalkene D reacts to form a pair of enantiomers (F & G) – draw these compounds in the appropriate boxes below (note – the labels F & G are arbitrary): (6 pt)



This question is continued on the next page...

(c) Cycloalkene A reacts with  $Br_2$  in  $CCl_4$  to form two different CHIRAL products (H & I) – draw these compounds in the appropriate boxes below (note – the labels H & I are arbitrary): (6 pt)



Two different CHIRAL compounds

(d) Circle ONE of the following words/phrases that best describes the relationship between compounds H & I: (3 pt)

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